

Planarians in Toxicology. Responses of Asexual *Dugesia dorocephala* to Selected Metals

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The planarian *Dugesia dorocephala* is a freshwater invertebrate found in unpolluted flowing surface waters. Planarians have a sensitive nervous system with synapses and a true brain (Best and Noel 1969) and evidence these in a variety of social and response behaviors (McConnell 1967). The inclusion of planarians in a screening battery would provide improved sensitivity in detecting toxicity because planarians commonly respond to lower levels of contamination than do other species (Burnham 1981; Barndt and Bohn 1985). Furthermore, planarians, but not daphnids, can be included in a higher tier since they may exhibit a multiplicity of complex responses including inhibition and stimulation of reproduction, teratogenesis and tumorigenesis (Best and Morita 1982; Schaeffer et al. 1991a).

Numerous toxicity tests have been conducted to determine the acute and chronic effects of toxicants to provide data necessary for the development of water quality criteria. However, water quality standards are generally based on single substance acute (48–96 hr) and subchronic/chronic (weeks to months) tests, and few studies have examined the toxicity of mixtures of pollutants using constituents at their regulated levels (Spehar and Fiandt 1986).

The appropriateness of Illinois water quality standards for metals was investigated using a 1-hr behavioral test based on the responses of the planarian *D. dorocephala*. One possible difficulty with water quality standards for metals is that the standard for each metal is usually established without regard to the effects of other metals present in the receiving water.

MATERIALS AND METHODS

Asexual *D. dorocephala* (18 to 20 mm; Carolina Biological Supply, Gladstone, Oregon) were maintained on an 8:16 hr light:dark cycle in medium containing 300 mg/L CaCl_2 , 120 mg/L MgSO_4 , 60 mg/L NaHCO_3 , and 6 mg/L KCl; hardness = 370 mg-equivalent CaCO_3 /L (Kostecky et al. 1989). Animals were fed 24–48 hr prior to the start of testing. Tests of metals were carried out at the 1987 Illinois water quality standards concentrations of: B, Fe, Ni, Se, and Zn = 1.0 mg/L; Cr(VI) = 0.05 mg/L; Cu = 0.02 mg/L; Pb = 0.1 mg/L. Some tests used levels 10 to 50 times higher. Mixtures containing each of these metals at 0.5, 1, 2, and 10 times the criteria levels were also

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tested. The pH was adjusted to 6 to 8 for these tests. Metal solutions were prepared in medium. Twenty animals were used at each test concentration. Independent duplicate trials were carried out.

Behavioral responses were recorded at 1, 2, 3, 4, 5, 10, 20, 30, 40, 50, and 60 min. Several distinct responses were scored, and the following were most frequently observed: (1) restlessness, (2) hyperkinesia, (3) spiraling, (4) head/nose twist, (5) shape change, and (6) labored movement. Animals often simultaneously showed several responses. Results were also evaluated using a modification of a scoring system devised for environmental samples using phenol as the reference toxicant (Grebe and Schaeffer 1991). Scores were evaluated relative to all the samples in the group. Data from other metals and organic compounds (Schaeffer et al. 1991b) provided an additional basis for evaluation. The scores used were:

Weak: A few animals show shape changes, nervousness, head/nose twist, or labored movement. The toxicity pattern was similar to that of animals to 32.6 mg/L phenol.

Moderate: Most animals are restless, and most show a variety of movement and shape-related responses. The pattern of toxic responses in these samples is similar to that produced by 65.3 mg/L phenol.

Strong: Most animals go through cycles of restlessness or hyperkinesia, labored movement increases with time and show concurrent shape changes, depression is evident, and some animals become morbid. In some instances, animals removed to clean medium die within 48 hours. Toxicity patterns are similar to exposure to 65.3 to 97.9 mg/L phenol.

To facilitate statistical analysis and interpretation, responses were grouped into 4 categories and 3 time groups: response category I = restlessness and hyperkinesia, II = spiraling and head/nose twist, III = other shape change, and IV = labored movement (Tables 1, 4) or relative to phenol (Table 3). Although other behavioral responses were found, these were observed only occasionally and were not considered in the analysis. These responses are from a response checklist for asexual *D. dorotocephala* (Grebe and Schaeffer 1991). A similar checklist of coded behavior and toxic signs is used for the fathead minnow (*Pimephales promelas*; Geiger et al. 1988).

The effects of concentration and time on the patterns of responses were determined using X^2 analysis with linear contrasts (Sauer and Williams 1989) as implemented in CONTRAST (Hines and Sauer 1989). Concentration effects were tested using count data transformed into failure rates (proportions) by dividing each count by the sum of all the counts. Temporal changes within a concentration were determined from failure rates obtained using the total number of counts for that concentration as the divisor.

RESULTS AND DISCUSSION

With the exception of Fe(III) ($P > 0.05$), the patterns of response for an individual metal significantly changed with time ($P < 0.05$) (Table 1). Generally, more animals showed classes III and IV responses as the duration of exposure increased. This suggests a concentration-response effect of the individual metal. Iron(II) is believed to be more easily absorbed across biological .awoff

Table 1. Behavioral responses in 20 planarians/concentration exposed to metals in medium of hardness of 370 mg-equivalent CaCO_3/L .

Metals ^b	1 to 5 min				p ^d	10 to 30 min				p ^d	40 to 60 min				p ^e	
	I ^a	II	III	IV		I	II	III	IV		I	II	III	IV		
B (1)	61	46	0	0	*	41	20	3	3	-	9	16	24	23	-	*
B (10)	72	66	1	0		32	30	2	1		14	26	9	8		*
Cr ⁶⁺ (0.05)	58	27	0	0	*	39	18	4	0	*	27	12	17	22	-	-
Cr ⁶⁺ (0.5)	110	49	17	0		14	23	5	0		29	16	26	23		*
Cu (0.02)	87	37	6	0	*	44	31	10	10	-	11	11	41	24	-	*
Cu (0.2)	105	38	26	0		39	23	28	6		12	7	49	44		*
Fe ₃ ⁺ (1)	115	66	0	0	*	41	23	5	0	*	40	17	4	0	-	*
Fe ₃ ⁺ (10)	35	13	0	0		17	15	0	0		85	34	14	4		*
Fe ₃ ⁺ (50)	98	77	17	0		45	12	2	0		29	12	12	14		*
Ni (1)	10	22	0	0		20	13	5	0	*	18	21	1	9	*	-
Ni (10)	13	64	9	0	*	24	15	3	0		20	16	4	5		*
Ni (50)	112	59	10	0		52	5	36	17		33	7	60	38		*
Pb (0.1)	87	9	10	5	*	37	8	6	6	-	16	10	19	20	*	*
Pb (1.0)	113	17	16	5		37	17	9	9		18	7	35	33		*
Se (1)	15	37	0	0	*	10	27	7	7	*	10	7	29	34	-	*
Se (10)	73	46	0	0		38	28	4	4		23	13	19	21		*
Zn (1)	13	23	4	0	*	9	10	1	0	*	7	16	1	7	*	*
Zn (10) ^c	131	91	73	0		45	41	43	23		11	4	48	36		*

^aRoman numerals I through IV represent clusters of responses. See text.

^bNumbers in parenthesis represent concentrations of metals in mg/L. Overall concentration effects (i.e., between rows) were significant at $P < 0.05$ for each metal. C40% mortality observed after 1 hr and 50% mortality after < 24 hr.

^dMeans between concentrations within times differ at $P < 0.05$ (*).

^eMeans for each time group within the concentration differ at $P < 0.05$ (*).

membranes than iron(III). This is true for the absorption of dietary iron through the mucosae in the small intestine. And since the animals were fasting, the uptake of fluid and solutes were mainly by cutaneous diffusion, so it is possible that the infiltration of Fe(III) ions may have been inhibited across the planarian skin in a similar manner. However, studies using other compounds of iron would be necessary to confirm this.

Each metal produced behavioral toxicity at its criterion concentration, and had overall response patterns which differed significantly ($P < 0.05$) with concentration. Depending on the metal, responses at different concentrations differed for all, or only some, time periods (Tables 1-3). For example, B and Cu concentrations differed only at 1-5 min whereas Ni and Zn concentrations differed in each time period. For each metal, both the absolute number of responses, and concentration differences, tended to diminish with time (Table 2). As judged by the total number of responses, nickel at 1-10 mg/L and zinc at 1 mg/L were the least toxic metals, but were the most toxic at 50 and 10 mg/L, respectively. Water hardness can affect the toxicity of Cu, Pb, and Zn (Spehar and Fiantdt 1986; USEPA 1986), so the hardness of the medium may account for some of the observed toxicity differences between metals.

Table 2. Total responses by time period and concentration for $n = 20$ planarians at each concentration.

Metal	Conc. mg/L	1-5 min	10-30 min	40-60 min	Total
B	1	107	67	72	246
B	10	139	65	57	261
Cr(VI)	0.05	85	61	78	224
Cr(VI)	0.50	176	42	94	312
Cu	0.02	130	95	87	312
Cu	0.20	169	96	112	377
Fe	1	181	69	61	311
Fe	10	48	32	137	217
Fe	50	192	59	67	318
Ni	1	32	38	49	119
Ni	10	86	42	45	173
Ni	50	181	110	138	429
Pb	0.10	111	57	65	233
Pb	1.0	151	72	93	316
Se	1	52	51	80	183
Se	10	119	74	76	269
Zn	1	40	20	31	91
Zn	10	295	152	99	546

Table 3. Phenol-equivalent toxicity rating of regulated metals (mg/L) and mixtures at 0.5x, 1x, 10x and 50x criterion level.

Metal	Weak	Moderate	Strong
Cu	---	---	0.02, 0.2
Cr(VI)	---	0.05	0.5
Fe(III)	1	10, 50	---
Ni	1, 10	10	50
Pb	---	---	0.1, 1.0
Se	---	1, 10	---
Zn	1	---	10
Mixture	0.5x	1x, 2x	10x

Mean overall response rates for the mixtures of 0.0240, 0.0450, 0.0435, 0.0542 responses/min, differed significantly between concentrations (0.5x, 1x, 2x, 10x criteria, respectively). With the exception of a nonsignificant difference between 1x and 2x, the severity of the responses increased significantly with concentration. Time periods I and III, but not II, differed between concentrations. Within each concentration except 2x, temporal effects were significant ($P < 0.05$).

Table 4: Behavioral responses in groups of 20 planarians exposed to metal mixtures at multiples of Illinois criterion levels (mg/L).

Mixture	Response ^a	Observation Time (Min)					
		1 to 5		10 to 30		40 to 60	
		I	II	I	II	I	II
0.5x Criterion		7	12	15	24	14	24
Criterion		27	51	21	35	18	28
2x Criterion		30	22	26	29	27	40
10x Criterion		62	41	27	19	33	35

^aResponse I = restlessness and hyperkinesia

Response II = spiraling and head/nose twist

The mixtures were separately compared with the integer-rounded mean counts for the individual metals at their criteria levels. For the 1x criterion mixture, the overall response rates for these two groups, and for period III differed marginally ($P < 0.07$), but did not differ for periods I and II ($P > 0.10$). However, the overall temporal subgroup response rates for the 0.5x mixture were less than the rates for the individual metals ($P < 0.05$). The overall response rate for the 2x mixture was significantly smaller than the rate for the individual metals ($P < 0.05$) and a significant temporal difference existed for period I ($P < 0.05$). The overall response rate for the 10x mixture, and for each subgroup, did not differ from the rates for the mean individual metals.

Toxicity generally increased with concentration and with exposure time.

These increases suggest that there is a potential for the metals, alone or in combination, to cause deleterious effects in exposed organisms at concentrations below present proposed single-chemical water criteria (USEPA 1986). Further study is therefore necessary to evaluate the interaction of chemical compounds and determine the effects of these on sensitive organisms. The rapid test used here provides a convenient bioassay for rapid, preliminary evaluation of water toxicity.

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